

**European Union / United Nations Economic Commission for Europe  
International Co-operative Programme on Assessment and  
Monitoring of Air Pollution Effects on Forests**

Meeting of the Working Group on QA/QC  
(subgroup of the Expert Panel Deposition)  
together with experts in chemical analysis from other Expert Panels

**Future work on QC/QA in laboratories:  
evaluation of data from monitoring programs under  
the aspect of QC/QA**

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- Continuing the assessment of the validation criteria, using interlaboratory studies (going on: ion balance and DOC in atmospheric deposition and soil water), involving different laboratories than those already participating in the WG.
- Pushing to get more laboratories performing all the chemical determination and using a QA/QC approach in their activity;
- Continuation on a regular basis (suggested yearly frequency) of the WRT for the analysis of leaves, soil, water.
- Working Ring Tests: is it possible to include common aspects to the three analytical fields (leaves, soil, water)?
- Creation of a space (regular meetings?) where persons working in analytics may have discussion.

## Aspects to be considered to assure good analytical quality of results

### **Field**

Sampling, transport and conservation of samples

### **Laboratory**

- Skilled personnel
- Validated and written analytical methods
- Properly constructed, equipped and maintained laboratory facilities
- Use of high-quality glassware, reagents, de-ionised water and other testing material

### Internal QC

- Calibration, adjustment, and maintenance of equipment
- Use of blanks, DL, QL
- Use of replicate samples
- Use of control samples and standard samples, with proper records (control charts)
- Validation and critique of results
- Archiving results

### External QC

- Interlaboratory exercises
- Certified reference materials

## Further steps in the validation of results:

### Comparison between measured (CM) and calculated conductivity (CE)

$$\mathbf{CD} = 100 * \frac{(CM-CE)}{CM}$$

for conductivity  $\leq 100 \mu\text{S cm}^{-1}$   $CE_{\infty} = \sum \lambda_i c_i$

for conductivity  $> 100 \mu\text{S cm}^{-1}$   $CE = \sum \lambda_i f_i c_i$

$\lambda_i$  *equivalent ionic conductance*

$C_i$  *Concentration of the ion i*

$f_i$  *activity coefficient*

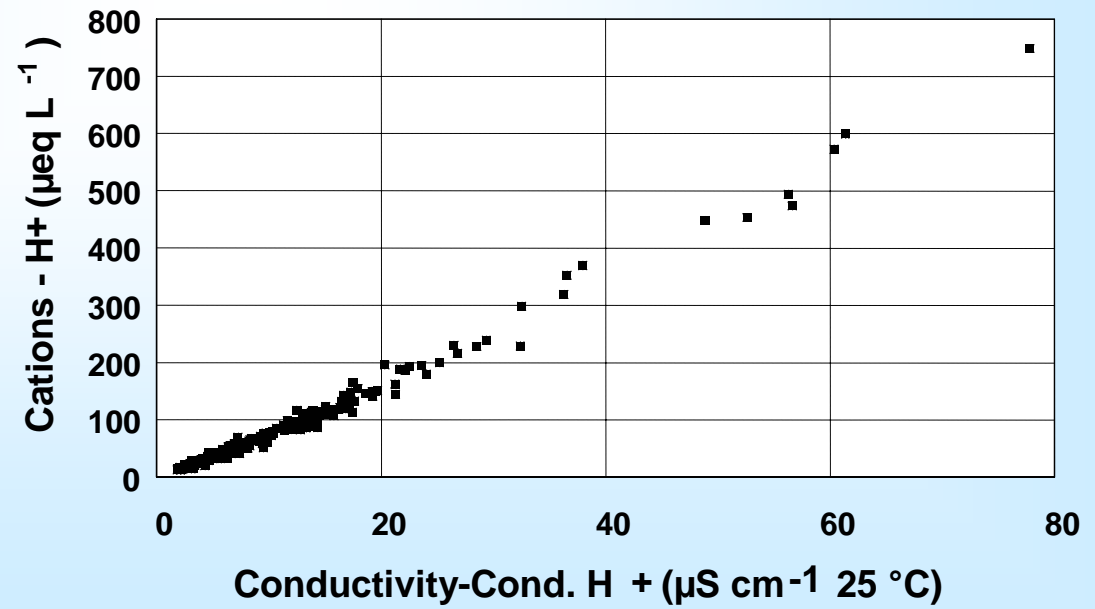
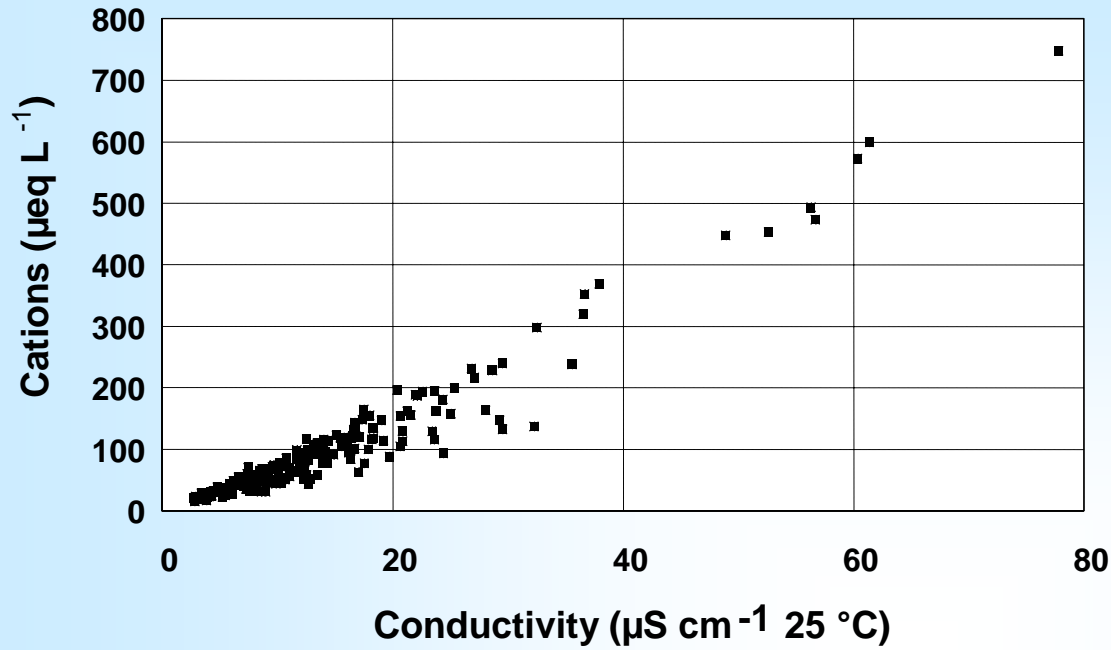
## Further steps in the validation of results:

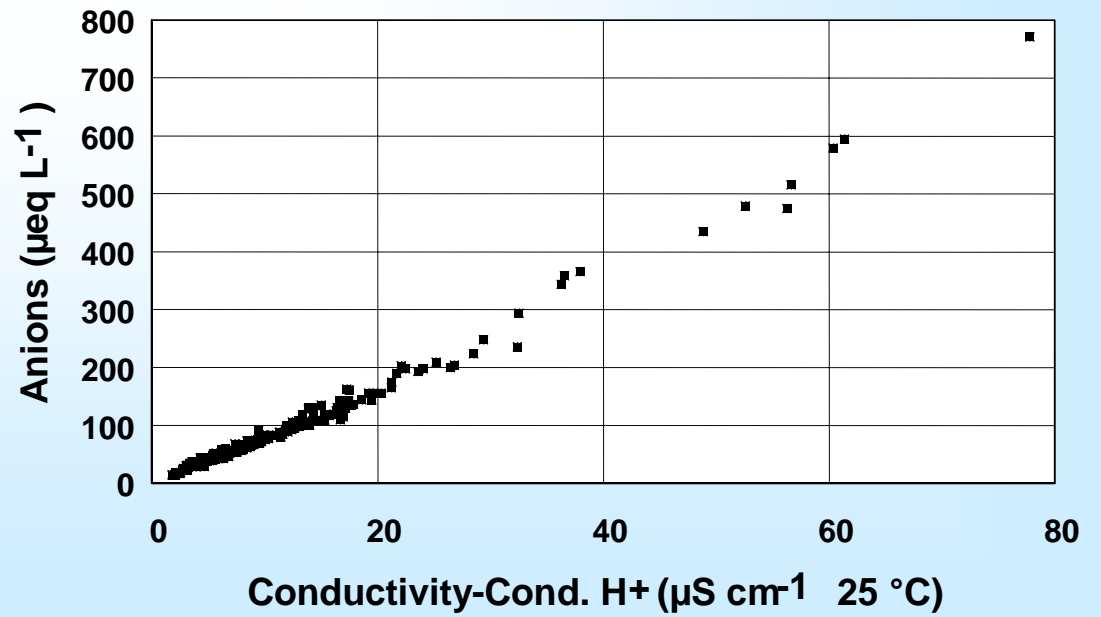
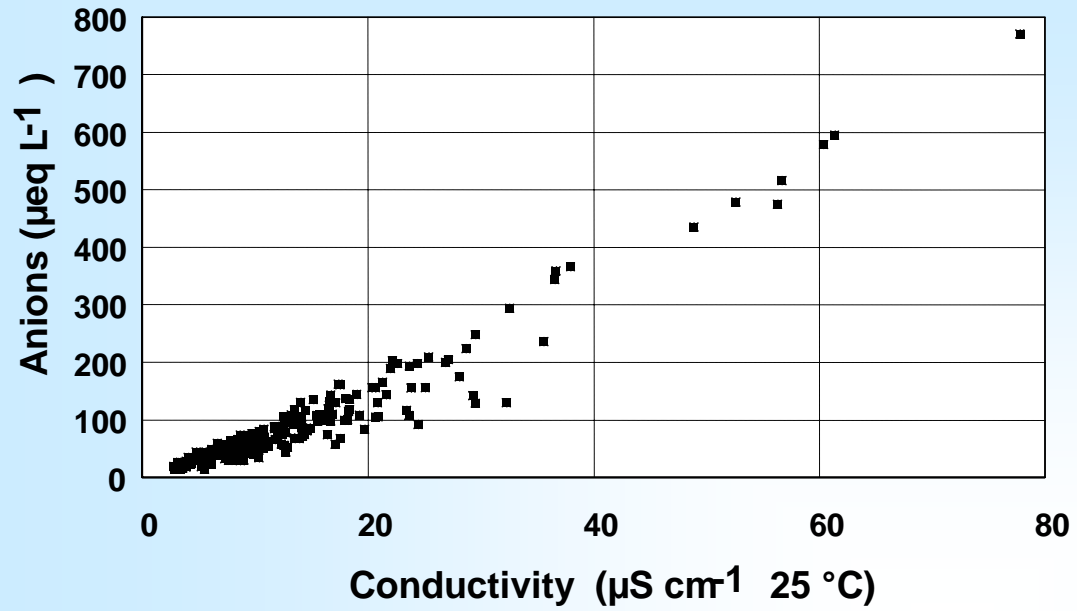
- Relationships between conductivity and ion (cation, anion) concentrations

It works nicely when hydrogen ion concentrations are low (pH>5.0);

When H<sup>+</sup> concentration is high, it contributes strongly to conductivity.

| Ions   | Equivalent conductance at 25°C<br>kS cm <sup>2</sup> eq <sup>-1</sup> |
|--|---|
| H <sup>+</sup>   | 0.350   |
| Ca <sup>++</sup> , Mg <sup>++</sup> , Na <sup>+</sup> , K <sup>+</sup> ,<br>NH <sub>4</sub> <sup>+</sup> , Cl <sup>-</sup> , SO <sub>4</sub> <sup>=</sup> , NO <sub>3</sub> <sup>-</sup> | 0.044-0.080   |

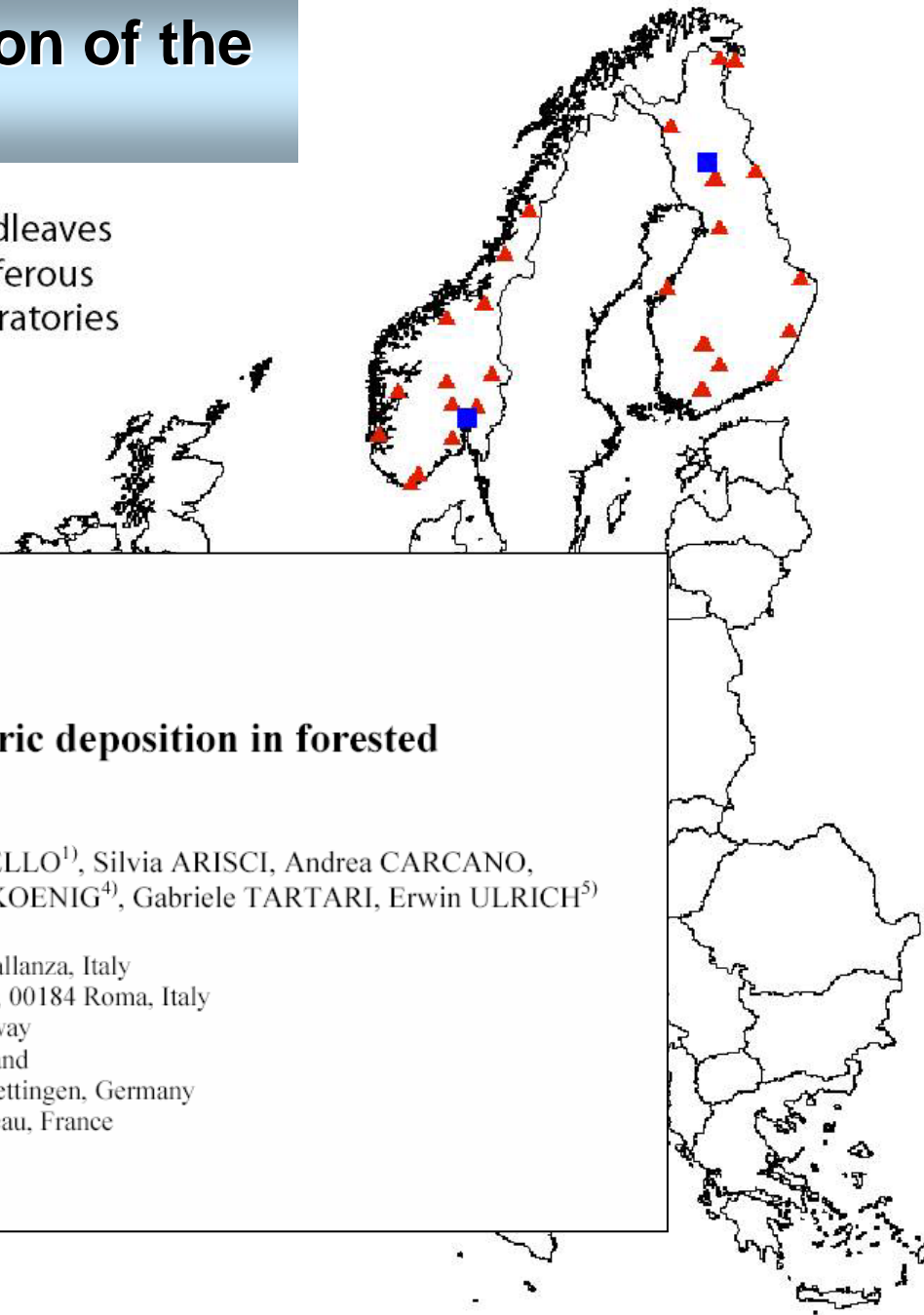




## Further steps in the application of the validation criteria

➤ About 5000 analyses of deposition samples done from 7 different laboratories

- Broadleaves
- ▲ Coniferous
- Laboratories



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### Validation of chemical analyses of atmospheric deposition in forested European sites

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# Criteria for the validation of the results of chemical analyses (atmospheric deposition, soil water)

## Second step

### Aims

Emphasise the use of data validation in the routine practice of analysis

Include DOC in the validation

Increase the number of laboratories involved in the exercise

Investigate on the meaning and relationships of DOC in atmospheric deposition and soil water

## Laboratories at present involved in the study

**Italy**

C.N.R. Institute of Ecosystem Study, Pallanza

**France**

SGS Laboratories Wolff-Environment, Evry

**Norway**

Norwegian Forest Research Institute, Ås

**Germany**

Niedersaechsische Forstliche Versuchsanstalt, Goettingen

**Switzerland**

WSL, Birmensdorf

**Denmark**

Forest & Landscape, Hørsholm

**Flanders (Belgium)**

Laboratorium Bodemkunde & IBW (pH and EC)

**UK**

Forest Research, Farhnam, Hampshire

# Ionic balance

$$\text{PD} = 100 * \frac{(\Sigma \text{ cat} - \Sigma \text{ an})}{0.5 (\Sigma \text{ cat} + \Sigma \text{ an})}$$

$$\Sigma_{\text{anions}} = \text{Alk} + [\text{SO}_4^=] + [\text{NO}_3^-] + [\text{Cl}^-] + [\text{Org}^-]$$

$$\Sigma_{\text{cations}} = [\text{Ca}^{++}] + [\text{Mg}^{++}] + [\text{Na}^+] + [\text{K}^+] + [\text{H}^+] + [\text{NH}_4^+]$$

[ Org- ] is measured as DOC (mg C L<sup>-1</sup>)

We indicate as DOC formal charge the apparent ionic charge of 1 mg/L of DOC assuming that:

- no errors are affecting the ion concentrations
- no other ions are present in solutions

## Ionic balance

$$\Sigma_{\text{anions}} = \text{Alk} + [\text{SO}_4^=] + [\text{NO}_3^-] + [\text{Cl}^-] + [\text{Org}^-]$$

$$\Sigma_{\text{anions}} = \text{Alk} + [\text{SO}_4^=] + [\text{NO}_3^-] + [\text{Cl}^-] + f(\text{DOC})$$

$$f(\text{DOC}) = \text{slope} * \text{DOC} + \text{intercept}$$

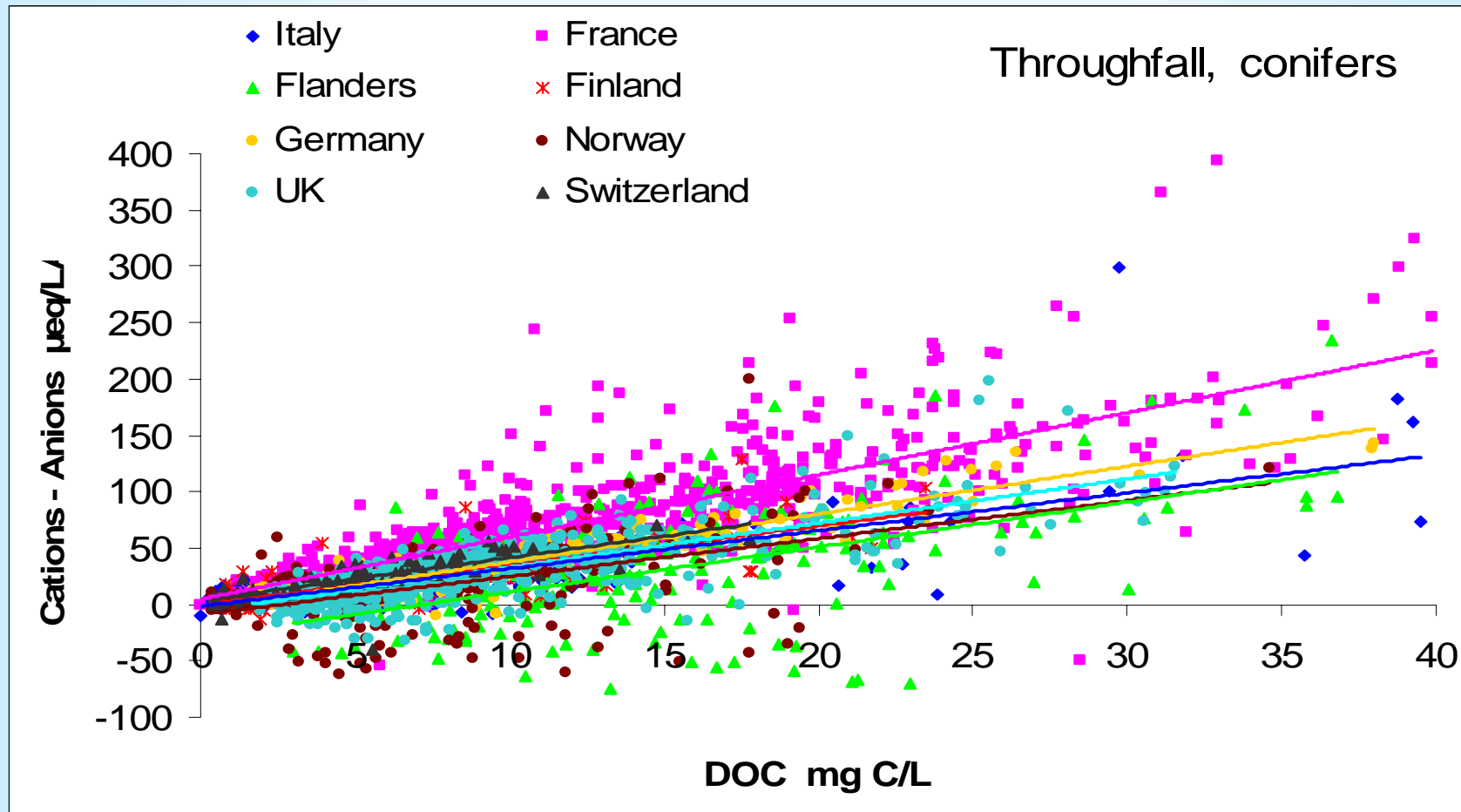
# Number of data collected

| Laboratory  | BOF | THR<br>Beech | THR<br>Oak | THR<br>other | THR<br>Broadleaves | THR<br>Pine | THR<br>Spruce | THR<br>other | THR<br>Conifers |
|-------------|-----|--------------|------------|--------------|--------------------|-------------|---------------|--------------|-----------------|
| Italy       | 659 | 278          | 231        | 0            | 509                | 0           | 110           | 59           | 169             |
| France      | 0   | 0            | 0          | 0            | <u>1361</u>        | 0           | 0             | 0            | 0               |
| Norway      | 181 | 0            | 0          | 0            | 0                  | 0           | 216           | 51           | 267             |
| Germany     | 268 | 99           | 33         | 0            | 132                | 66          | 198           | 0            | 264             |
| Switzerland | 307 | 132          | 88         | 0            | 220                | 45          | 42            | 28           | 115             |
| Denmark     | 101 | 130          | 71         | 0            | 201                | 0           | 195           | 0            | 195             |
| Flanders    | 402 | 208          | 0          | 105          | 313                | 108         | 0             | 0            | 108             |
| UK          | 375 | 0            | 307        | 0            | 307                | 848         | 0             | 0            | 848             |

| Laboratory  | STF<br>Broadleaves | STF<br>Conifers | Soil water<br>metals           |
|-------------|--------------------|-----------------|--------------------------------|
| Italy       | 125                | 0               | 0                              |
| France      | <u>194</u>         | 0               | 1562<br>No data                |
| Norway      | 0                  | 0               | 267 *<br>T Al, Fe, Mn          |
| Germany     | 0                  | 0               | 1416 *<br>T Al, Al org, Fe, Mn |
| Switzerland | 0                  | 0               | 0                              |
| Denmark     | 0                  | 0               | 798 *<br>T Al, Fe, Mn          |
| Flanders    | 105                | 0               | 475 *<br>T Al, Fe              |
| UK          | 0                  | 0               | 1287 *<br>T Al, Fe, Mn         |

- Contribution of the organic carbon to the ion balance

➤ The relationship between the difference  $\sum \text{cations} - \sum \text{anions}$  and the DOC concentration is tested. On this base a formal charge per mg/liter of organic carbon is assigned.



The slope of the regression DOC (mg L<sup>-1</sup>) vs Cat-An (μeq L<sup>-1</sup>)  
is an evaluation of the DOC Formal Charge

$$\text{Slope} = \Delta y / \Delta x = \mu \text{ eq L}^{-1} / \text{mg C L}^{-1} = \mu \text{eq} / \text{mg C}$$

## Relationship between ( $\sum \text{cat} - \sum \text{an}$ ) and DOC

| Country     | slope | intercept | R <sup>2</sup> |
|-------------|-------|-----------|----------------|
| Italy       | 3,35  | -1,53     | 0,57           |
| France      | 5,52  | 4,62      | 0,65           |
| Flanders    | 3,99  | -28,79    | 0,29           |
| Finland     | 3,51  | -0,39     | 0,60           |
| Germany     | 4,23  | -4,43     | 0,85           |
| Norway      | 3,37  | -8,77     | 0,24           |
| UK          | 3,91  | -6,08     | 0,62           |
| Switzerland | 3,79  | 3,70      | 0,61           |



## Data treatment

- Data of each laboratory were validated using the standard excel file for the data validation, available in the web. Data of each laboratory were aggregated on the basis of (1) each single plot and (2) of the type of vegetation.
- Other graphs were added to those already present in the validation file. They are useful for a general exploration of data and include the relationship between **DOC** and  $\Sigma$  **cations** –  $\Sigma$  **anions**.
- The data used for the evaluation of DOC Formal Charge include those not fitting the validation criteria, but do not include the highest values (strong skewness).
- The evaluation of DOC FC using the slope of the linear regression must be considered as a preliminary approach.

## General comments

✓ This approach requires high precision in the analyses, as the difference ( $\sum \text{cat} - \sum \text{anions}$ ) cumulate the systematic and random errors performed in the determination of each ion. In particular it is strongly dependent on systematic errors.

✓ Is it possible to find values of “formal charge” per mg/L of TOC for different types of solutions (bulk open field, throughfall, stemflow, soil water)?

✓ This can be reached with a statistical approach of data set obtained in different laboratories, identifying likely relevant variables:

Type of solution (bulk open field, throughfall, soil water, etc.)

Type of vegetation

Yearly amount of precipitation

Mean air temperature

??

## Soil water

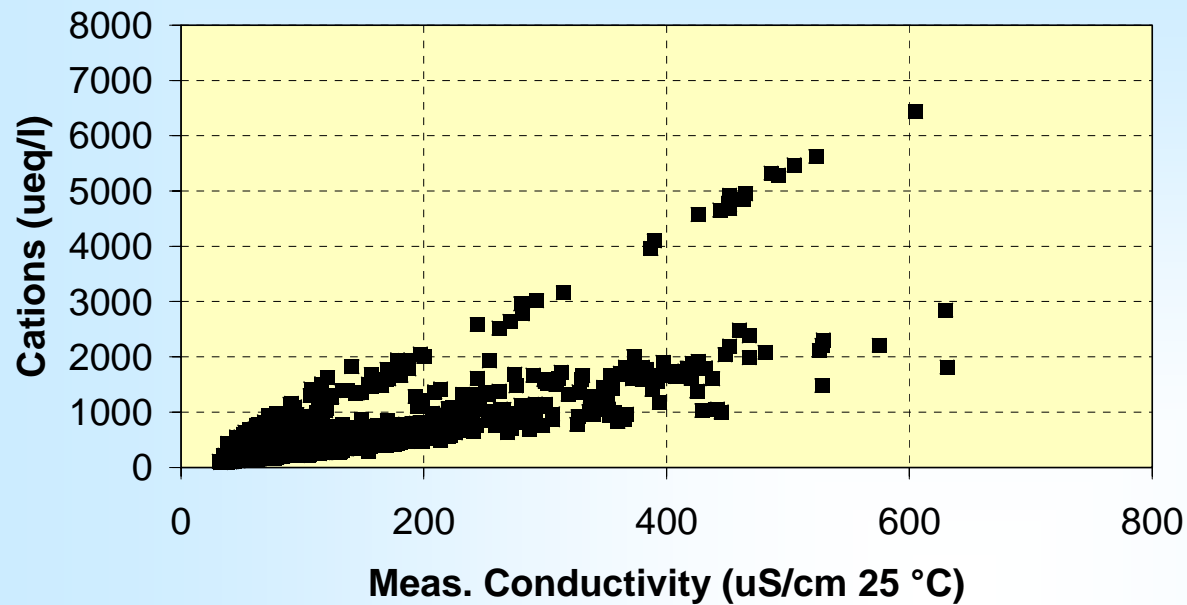
In the case of soil water both organic carbon and trace metals are important in the ion balance.

However there are problems in evaluating the ionic contribution of trace metals to the ion balance, because of the dependence of the metal speciation from pH and the possibility of complexation with organic substance (DOC).

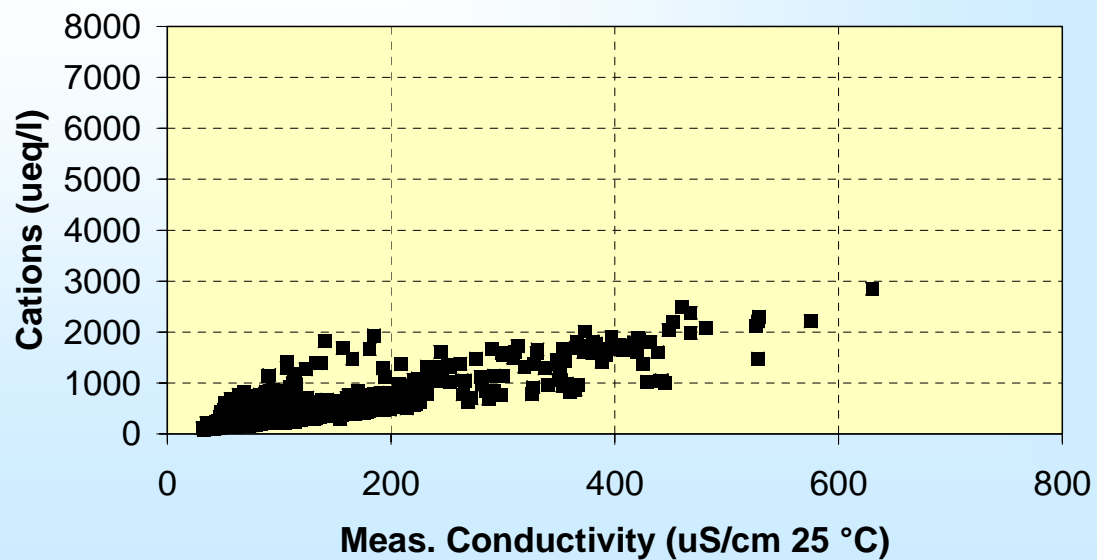
These aspects do not exclude the possibility to check for correlations among variables, assuming a full dissociation of trace metals

**all samples**

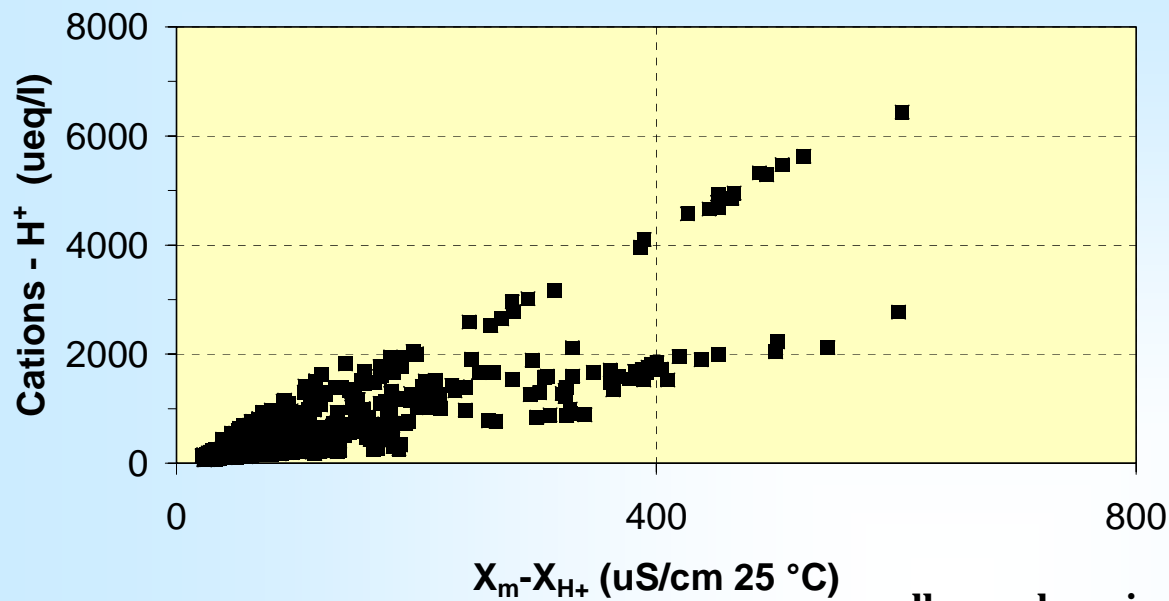
Germany, soil water



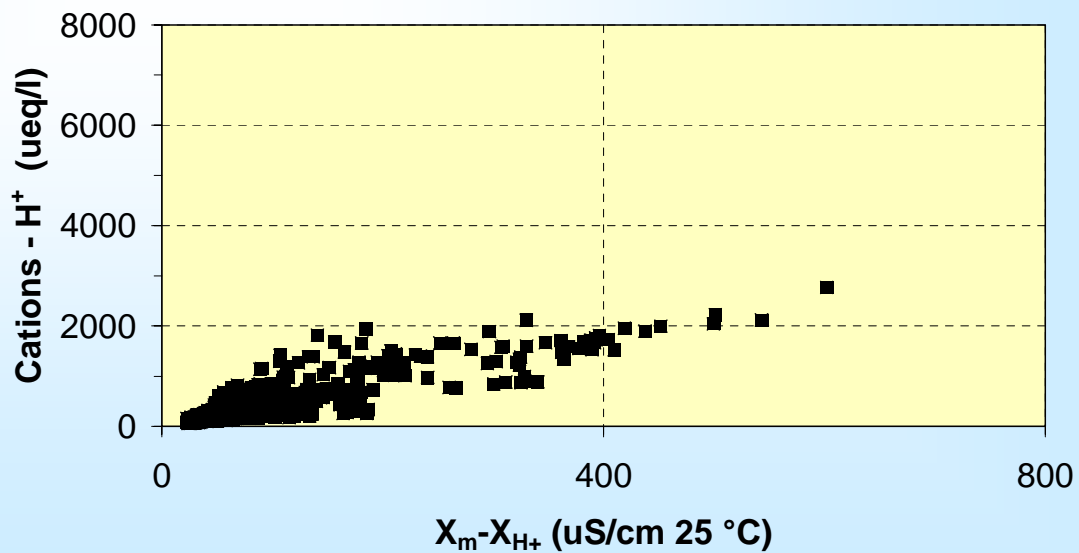
**all samples minus samples without Al (Al>1 mg/l)**



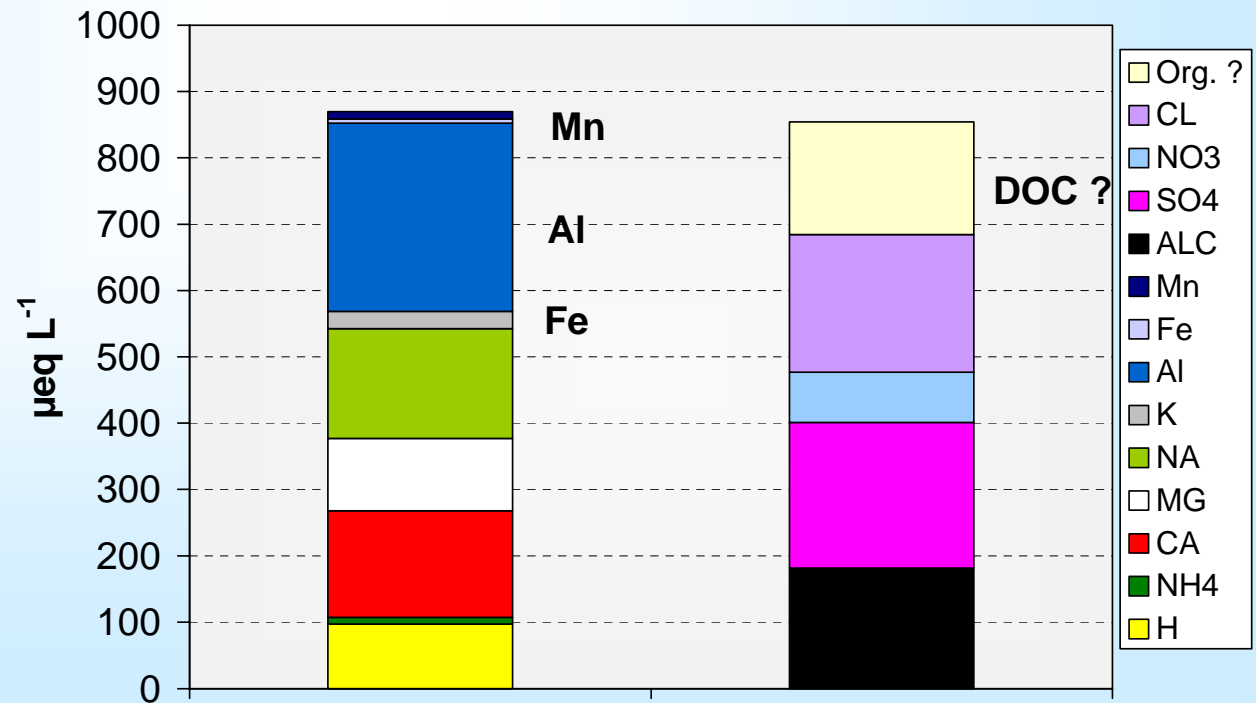
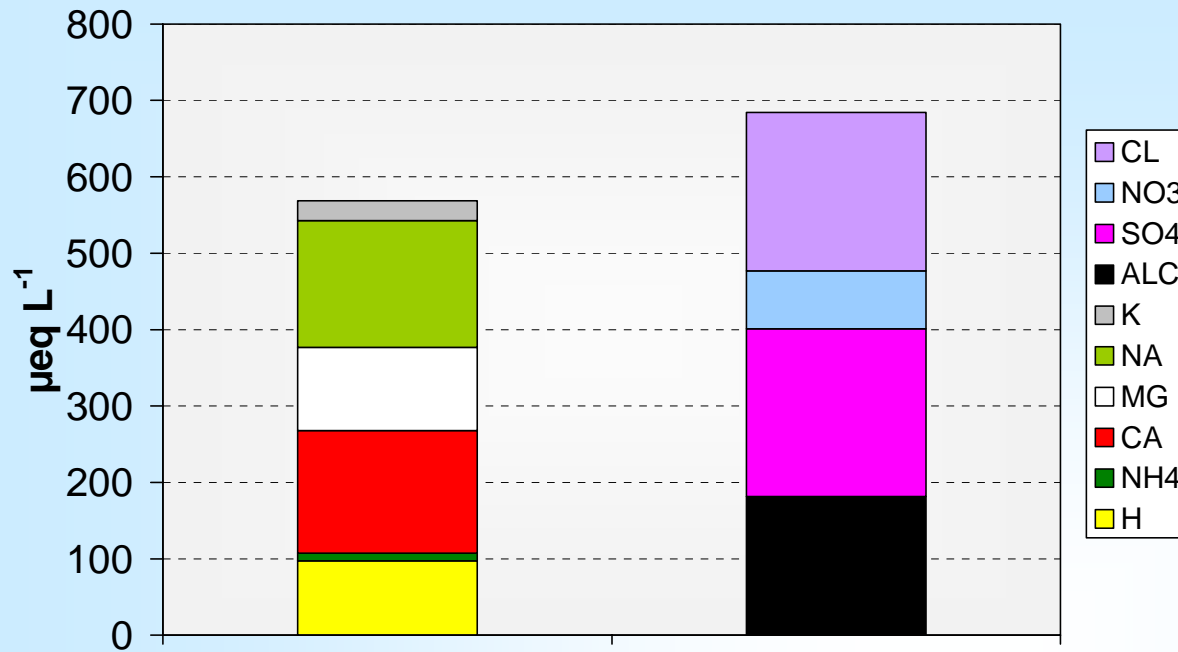
**all samples**



**all samples minus samples without Al (> 1 mg/l)**



# Germany, soil water



## Partial conclusions

Four different tests for the validation of chemical analyses were identified; they help in the data screening, but they should be used with care, taking into account their limits.

An excel file makes easier the use of such criteria.

These criteria are exactly the same indicated in the ICP Forests manual, Part VI, Sampling and analysis of Deposition.

Both the manual and the excel file are easily downloadable from the ICP Forests web page <http://www.icp-forests.org/>.

Additional techniques for the data validations are under evaluation.

Interlaboratory studies are useful tools to increase awareness of the need of a QA/QC approach to analytical activity.